

CK5

not on checklist

PTS

December 2nd

1. Born-Haber/ lattice energy
2. Gases
3. liquids
4. solids

12

34

27

17

90

fcc

389 pm

atomic radius

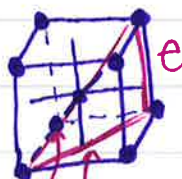
simple



* $e = 2r$

* 1 atom/unit cell

body-centered



* $4r = \sqrt{3}e$

* 2 atoms/unit cell

face-centered



* $4r = \sqrt{2}e$

* 4 atoms/unit cell

↳ $8 * \frac{1}{8}$ corners

↳ $6 * \frac{1}{2}$ face centered

big
4 ions/unit cell

* big ions = usually anions

1 big ion/unit cell

2 big ions/unit cell

$v = e^3$

density = $\frac{m}{v}$

* of one unit cell

↳ $e = 2r$

1 * AM/N_A simple

2 * AM/N_A bcc

4 * AM/N_A fcc

(of one unit cell)

→ octahedral

hole: space surrounded by 6 atoms

↳ from FCC

→ tetrahedral hole:

↳ 8 holes in every fcc unit cell

* closest way to pack stuff!

1.55 g/L @ 300K 99.4 kPa
 400K 107 kPa

December 2nd

$$\frac{P_1 V_1}{n_1 T_1} = \frac{P_2 V_2}{n_2 T_2}$$

$$\frac{P V M M}{M T} = \frac{P V M M}{M T}$$

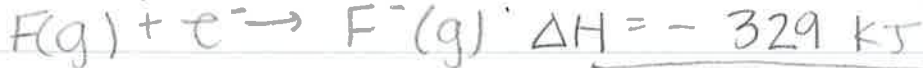
$$\frac{P M M}{d T} = \frac{P M M}{d T}$$

	①	②
P	99.4 kPa	107 kPa
M		
d	1.55	d ₂
T	300K	400K

p. 225 (in textbook)



$$\Delta H = +89 \text{ kJ/mol}$$



FCC (Ionic)

4 large ions/unit cell

8 tetrahedral holes

$$\hookrightarrow (8 * \frac{1}{8} + 8 * \frac{1}{2})$$

4 oct. holes/unit cell

fcc Na₂S

\hookrightarrow 12 edge-centers,
each 1/4 in = 3

+1 in body-centered position