

# Dr. Mattson's CHM 203!

8/26/2015 Wednesday

Review Sessions with Monika Lukauskas

Thursdays + Sundays 7-8:30 pm

Room → TBA

Today's lesson: Sections 1.1-1.5  
~~52(a)~~ → 52(c)

Section 1 → scientific method → Read it!  
It will be on the test!

Section 2 → measurement → read it to know what's going on

<u>SI quantities</u>	
Mass	kg
Length	m
Temp	K
amount	mol
time	s

$10^{12}$	Tera	T
$10^9$	Giga	G
$10^6$	Mega	M
$10^3$	Kilo	k
$10^{-2}$	Centi	c)
$10^{-3}$	milli	m
$10^{-6}$	micro	$\mu$
$10^{-9}$	nano	n
$10^{-12}$	pico	p
$10^{-15}$	femto	f
$10^{-18}$	atto	a

relationships

①  $1 \text{ g} = 1 \text{ g}$

②  $1 \text{ mg} = \frac{1 \times 10^{-3}}{1} \text{ g}$

**MISTAKE:**  
 $1 \times 10^{-9} \text{ ng} = 1 \text{ g}$

Example

Which is larger?

15 mg or  $4.2 \times 10^{-2} \text{ g}$

to solve, change g  $\rightarrow$  mg  
quantity

mass =  $4.2 \times 10^{-2} \text{ g} \times \frac{1 \text{ mg}}{10^{-3} \text{ g}}$  = 42 mg  
 value + units

①  $1 \text{ mg} = 10^{-3} \text{ g}$

②  $\frac{1 \text{ mg}}{10^{-3} \text{ g}} = 1$

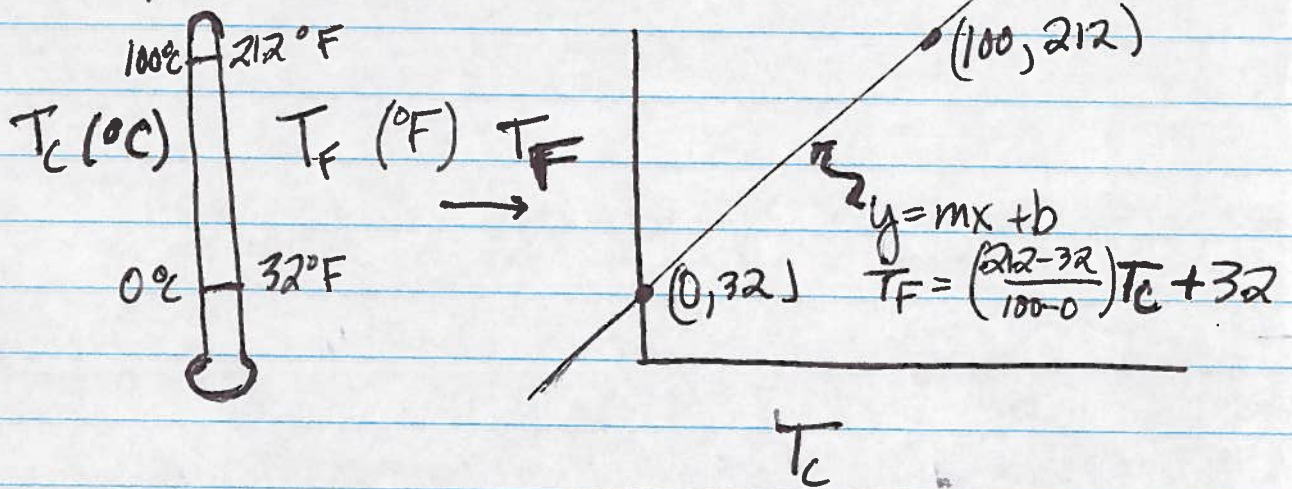
Convert 157 pm into nm.

$$\text{Length} = 157 \text{ pm} \times \frac{10^{-12} \text{ m}}{1 \text{ pm}} \times \frac{1 \text{ nm}}{10^{-9} \text{ m}} = .157 \text{ nm}$$

①  $1 \text{ pm} = 1 \times 10^{-12} \text{ m}$   
 $1 \text{ nm} = 1 \times 10^{-9} \text{ m}$

②  $1 = \frac{10^{-12} \text{ m}}{1 \text{ pm}} \times \frac{1 \text{ nm}}{10^{-9} \text{ m}} = 1$

## Temperature



$$\underline{T_F = 1.8 T_C + 32} \approx \text{pg 10} \quad \underline{^\circ\text{F} = 1.8^\circ\text{C} + 32^\circ\text{F}}$$

Kelvin

$$\underline{T_K = T_C + 273}$$

$$T_F \leftrightarrow T_C \leftrightarrow T_K$$

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1. Classify each of the following as a unit of mass, length, or temperature.

- a. mg mass      b. dL length<sup>3</sup>/volume      c. K temp  
 d.  $\mu\text{L}$  length<sup>3</sup>/volume      e. pm length      f. nm length

2. Complete these equalities:

a.  $1 \text{ mm} = \underline{10^{-3}} \text{ m}$

f.  $1 \text{ kg} = \underline{10^3} \text{ g}$

b.  $1 \text{ mL} = \underline{10^{-3}} \text{ L}$

g.  $1 \text{ pm} = \underline{10^{-12}} \text{ m}$

c.  $1 \text{ mg} = \underline{10^{-3}} \text{ g}$

h.  $1 \mu\text{L} = \underline{10^{-6}} \text{ L}$

d.  $1 \mu\text{m} = \underline{10^{-6}} \text{ m}$

i.  $1 \text{ cm} = \underline{10^{-2}} \text{ m}$

e.  $1 \text{ ng} = \underline{10^{-9}} \text{ g}$

j.  $1 \text{ mm} = \underline{10^{-3}} \text{ m}$

3. The radius of a potassium atom is 280 pm. Convert this to units of nm. This is a unit of length. Start by identifying the desired quantity, as we did in class:

$$\frac{\text{length}}{\text{(Quantity)}} = \frac{280 \text{ pm}}{1} \left| \frac{10^{-12} \text{ m}}{1 \text{ pm}} \right| \left| \frac{10^9 \text{ nm}}{1 \text{ m}} \right| = \underline{0.280 \text{ nm}}$$

(value) + (units)

$$\rightarrow \approx \frac{1 \text{ nm}}{10^{-9} \text{ m}}$$

4. Rearrange the following equation to solve for x:  $4y = 2x + 7$ 

$$-2x = -4y + 7$$

$$x = 2y + \frac{7}{2}$$

5. Rearrange the following equation to solve for x:  $4y = 2(x + 7)$ 

$$4y = 2x + 14$$

$$x = 2y - 7$$

$$-2x = -4y + 14$$

Check in the any problem that you found difficult (you wouldn't be able to solve it on your own on an exam).