

Today 8/27 Finish chapter 1

↳ finish all chapter 1 worksheets and book problems.

Tuesday 8/28 - bring lab manual (no need to read expt before lab)

- lab notebook

- laptop computer

- write intro in lab notebook

- safety glasses not required tomorrow

Tuesday evening: Problem club with Ali 7:00 - 8:30 HUSB 359

1.94b

nutritional calories, 1 Cal = 1000 calories, 1 cal = 4.184 J

$$\frac{540 \text{ Calories} \left| \frac{1000 \text{ kcal}}{1 \text{ Cal}} \right| \frac{4.184 \text{ kJ}}{1 \text{ kcal}}}{1 \text{ Cal} \quad 1 \text{ kcal}} = \boxed{2.26 \times 10^3 \text{ kJ}}$$

1 Watt = 1 J/s

$$\frac{2.26 \times 10^3 \text{ kJ} \left| \frac{1000 \text{ J}}{1 \text{ kJ}} \right| \frac{1 \text{ watt}}{1 \text{ J/s}} \left| \frac{1 \text{ min}}{60 \text{ s}} \right| \frac{1 \text{ hr}}{60 \text{ min}} \left| \frac{1}{100 \text{ watt}} \right|}{1 \text{ watt}} = \boxed{6.3 \text{ hours}}$$

1.52b

convert $8.5 \text{ cm}^3 \rightarrow \text{m}^3$

$$\frac{8.5 \text{ cm}^3 \left| \frac{(1 \times 10^{-2})^3 \text{ m}^3}{1^3 \text{ cm}^3} \right| \frac{1^3 \text{ mm}^3}{(1 \times 10^{-3})^3 \text{ m}^3}}{1^3 \text{ cm}^3} = \boxed{8.5 \times 10^3 \text{ mm}^3}$$

Aluminum has a density of 2.70 g/cm^3 . How thick is a 12 in x 18 in sheet of Aluminum with a mass of 5.3g?

$$\frac{5.3 \text{ g}}{2.70 \text{ g/cm}^3} = 1.96 \text{ cm}^3 = \left(12 \text{ in} \left| \frac{2.54 \text{ cm}}{1 \text{ in}} \right.\right) \times \left(18 \text{ in} \left| \frac{2.54 \text{ cm}}{1 \text{ in}} \right.\right) \times \text{thickness}$$

$$\boxed{\text{thickness} = 0.0014 \text{ cm}}$$

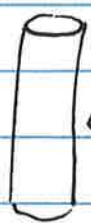
Problems from day 2 worksheet

(4) $d_{\text{air}} = 1.21 \text{ g/L}$ $d = \frac{m}{V} \Rightarrow m = d \cdot V$ $1 \text{ mL} = 1 \text{ cm}^3$

$$m = d \cdot V = \frac{1.21 \text{ g}}{\cancel{\text{L}}} \cdot \frac{1 \text{ m}^3}{1 \text{ mL}} \cdot \frac{1 \times 10^{-3} \cancel{\text{L}}}{1 \text{ cm}^3} \cdot \frac{1 \text{ mL}}{1 \text{ cm}^3} \cdot \frac{(1)^3 \text{ cm}^3}{(1 \times 10^{-2})^3 \text{ m}^3} = 1.21 \times 10^3 \text{ g}$$

(5) $E_k = \frac{1}{2} m v^2$ $1 \text{ J} = \frac{1 \text{ kg m}^2}{\text{s}^2}$ $m = 145 \text{ g}$

$$E_k = \frac{1}{2} \cdot \frac{145 \text{ g}}{1 \text{ g}} \cdot \frac{1 \times 10^{-3} \text{ kg}}{1 \text{ g}} \cdot \frac{(40 \text{ m/s})^2}{(\text{s})^2} \cdot \frac{1 \text{ J} \cdot \text{s}^2}{1 \text{ kg m}^2} \cdot \frac{1 \times 10^{-3} \text{ J}}{1 \text{ J}} = .116 \text{ kJ}$$

(8)  diameter = $\frac{0.010 \text{ in}}{1 \text{ in}} \cdot 2.54 \text{ cm} = 2.54 \times 10^{-2} \text{ cm}$
 $h = 1.0 \text{ m} = 100 \text{ cm}$
 $A = \pi r^2 = \pi \left(\frac{2.54 \times 10^{-2} \text{ cm}}{2} \right)^2$
 $A = 5.07 \times 10^{-4} \text{ cm}^2$
 $V = (5.07 \times 10^{-4} \text{ cm}^2) (100 \text{ cm})$
 $V = 5.07 \times 10^{-2} \text{ cm}^3$
 $d = \frac{0.2722 \text{ g}}{5.07 \times 10^{-2} \text{ cm}^3} = \boxed{5.37 \text{ g/cm}^3}$